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Biodegradation and decolorization of azo dyes by adherent *Staphylococcus lentus* strain

Kamel Chaieb¹ · Mohamed Hagar^{2,3} · Nagi R. E. Radwan^{2,4}

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Abstract A *Staphylococcus lentus* strain, isolated from Red sea water, was tested for decolorization capacity of Congo red, Evans blue, and Eriochrome Black T azo dyes. Biodegradation (100 mgl⁻¹) of these dyes was studied within 24 h in Mineral Salt Medium solution containing 0.10 % (w/v) yeast extract and 7 mM of glucose at a pH of 7.2 and a temperature of 37 °C. Using a 2.2×10^6 CFU/ mL inoculum size, *S. lentus* was able to decolorize these azo dyes with different degree. The staphylococcal biomass achieves approximately 100 % decolorization of Congo red and Eriochrome Black T by the end of treatment. FTIR and UV–Vis analyses confirm biodegradation potential of the tested strain. Moreover, the phytotoxicity of the dye solutions resulting from this treatment shows lower toxic nature compared to untreated solution of the respective dyes.

Keywords Azo dyes · Biofilm · Decolorization · *Staphylococcus lentus* · Phytotoxicity

Kamel Chaieb chaieb_mo@yahoo.fr

- ¹ Biology Department, College of Sciences, Taibah University, Yanbu el Baher, Kingdom of Saudi Arabia
- ² Chemistry Department, College of Sciences, Taibah University, Yanbu el Baher, Kingdom of Saudi Arabia
- ³ Chemistry Department, Faculty of Science, Alexandria University, Alexandria, Egypt
- ⁴ Chemistry Department, Faculty of Science, Suez University, Suez, Egypt

Introduction

Environmental pollution is one of the most important problems of the modern world. Synthetic dyes are used in paper printing, leather, and textile processing industries (Tony et al. 2009). Azo dyes are widely used due to their covalent attachment to the fibers, brilliant colors, and minimal energy consumption. The textile industry discharges a large amount of colored wastewater causing a considerable environmental pollution (Sanmuga Priya et al. 2015).

These dyes are a diverse class of chemicals containing various sulfonate groups and other substituents which affect its rate of decolorization (de los Cobos-Vasconcelos et al. 2012). A number of conventional physico-chemical wastewater treatment processes such as chemical, electro-coagulation, adsorption, flocculation, ozonation, and reverse osmosis have been proposed for decolorization of colored effluent (Robinson et al. 2001; Jadhav and Govindwar 2006). However, various synthetic dyes may escape from these treatment processes and persist in the environment for a long time.

Biodegradation is a process by which different types of microorganisms use complex chemical molecule and convert them into a simpler one (Solís et al. 2012). Biological treatment of effluent using fungi (Parshetti et al. 2007), bacteria (Kalyani et al. 2009), and yeast (Waghmode et al. 2011) has been reported elsewhere. Ding et al. (2014) noticed that *Sphingomonas* had a high biodegradation potential against azo dye in anaerobic systems. In addition, Reactive Red 2 was shown to be degraded by *Pseudomonas* sp. cells (Kalyani et al. 2009). Recently, Cui et al. (2014) reported that *Klebsiella* sp. Y3 was capable to decolorize azo dyes (Methyl Red, Congo Red, Orange I, and Methyl Orange) under anaerobic conditions.

Several microorganisms have developed enzyme systems for the efficient decolorization and metabolization of azo dye to colorless aromatic amines using azoreductases under both aerobic and anaerobic conditions, including *Clostridium* sp. (Rafii et al. 1990), *Enterobacter agglomerans* (Moutaouakkil et al. 2002), *Pseudomonas aeruginosa* (Nachiyar and Rajakumar 2005), *Sphingomonas paucimobilis* (Ayed et al. 2011), *Staphylococcus epidermidis* (Ayed et al. 2010), *Staphylococcus arlettae* (Elisangela et al. 2009), *Staphylococcus gallinarum* (Chen et al. 2011), *Staphylococcus cohnii* (Yan et al. 2012), *Dermacoccus abyssi* MT1 (Lang et al. 2014), *Acinetobacter baumannii* YNWH 226 (Ning et al. 2014), and *Staphylococcus aureus* (Pan et al. 2015).

In this study, *S. lentus*, isolated from Red sea water, was tested for its ability to decolorize three azo dyes (Congo red, Evans blue, and Eriochrome Black T). In addition, the phytotoxicity of the dye solutions before and after treatment was assessed.

Materials and methods

Dyes and chemicals

Congo red, Evans Blue, and Eriochrome Black T (Table 1) were purchased from Sigma (Chemical Company, MO, U.S.A.).

Sample collection, bacteria isolation, and culture conditions

Fresh water sample (100 mL) was collected in a sterile glass container from Red Sea Coast, Yanbu el Bahr, El Madina Elmounaoura, KSA.

Water sample (25 mL) was added to halophilic broth (Microgen, Maharashtra, India) containing the respective dyes (100 mgl⁻¹) and incubated at 37 °C. After 2 days of incubation, 1 mL of the culture was serially diluted and spread on halophilic agar (Microgen) plates containing 100 mgl⁻¹ of dyes. Then, colonies were screened for their ability to form a clear zone around the respective dyes (Ayed et al. 2009). Bacteria with positive reaction were isolated and subcultured in halophilic broth containing the respective dyes (100 mgl⁻¹). Suspected colonies of *Staphylococci* were identified by their positive Gram stain, and catalase positive reaction (Oxoid, Basingstoke, UK). Species identification was performed using API 20 Staph strips (Bio-Merieux, France) according to the manufacturer's recommendation. The microorganism responsible for a significant degradation of the tested dye was found to be S. lentus.

Adherence assay to measure biofilm production by *S. lentus* cells

Semi-quantitative adherence assay was performed on 96-well plates as described previously (Christensen et al.

Table 1 Molecular formula and chemical structure of dyes used in this study

Dyes Molecular formula Structure Chemical class $\lambda_{\rm max}$ (nm) NH_2 Congo red $C_{32}H_{22}N_6Na_2O_6S_2$ NH₂ 488 Azo dye 0=\$=0 O=S=O ÓΝa ÓNa Evans blue 0 C34H24N6Na4O14S4 611 Azo dye NH₂ OH OH NH2 NaC ℃H₃ H_o(0=\$=0 0=S=0ÓNa ÓNa Eriochrome black T C20H12N3NaO7S ONa 503 Azo dye $O=\dot{S}=O$ O_2N OH OH N=Ń

1985). An overnight culture of S. lentus was diluted to 1:100 in Tryptic Soy Broth supplemented with 2 % (w/v) glucose. A total of 200 µL of each cell suspension was transferred in a U-bottomed 96-well microtiter plate and incubated for 24 h at 37 °C. After incubation, bacterial culture was removed, washed three times with phosphatebuffered saline, and then dried in an inverted position. Adherent bacteria were fixed with 95 % ethanol and were colored with 100 µl of 1 % (w/v) crystal violet solution for 5 min. The wells were rinsed three times with sterile distilled water and air dried for 2 h. The optical density (OD) of each well was measured at 570 nm using an automated Multiskan reader (ELX800, Biotek instrument, USA). Biofilm formation was interpreted as follows: highly positive (OD₅₇₀ \geq 1), low-grade positive (0.1 \leq OD₅₇₀ < 1), or negative $(OD_{570} < 0.1)$ (Knobloch et al. 2001).

Organisms and culture conditions for biodegradation assay

Dyes were prepared in MSM, filter sterilized through 0.22 μ m filter, and stored at +4 °C in the dark until use. Glucose (7 mM) and yeast extract (0.1 % w/v) was prepared in MSM and sterilized.

The pH of MSM was adjusted to 7.2, supplemented with 7 mM of glucose, 0.1 % (w/v) of yeast extract, 1 mL of 2.2×10^6 CFU/mL, and 100 mgl⁻¹ of azo dyes (agitation: 150 rpm, T: 37 °C, and pH: 7.2). The yeast extract, glucose, and dyes were added to sterilize MSM.

Color removal was determined spectrophotometrically (UV-1800 SHIMADZU, Japan) based on the maximum absorbance of UV–Visible spectrum (Jadhav et al. 2008) using the following equation:

Color removal (%) = $(A_i - A_t) \times 100 / A_i$,

where A_i is the initial absorbance and A_t is the absorbance at incubation time t.

The efficiency of dye removal was expressed as the percentage ratio of decolorized dye concentration to that of the initial one.

Fourier transform infrared spectroscopy

The FTIR spectra of the samples before and after decolorisation were done using Fourier transform infrared spectroscopy (Nicolet iS 10 Thermoscientific). The samples were dried and mixed with spectroscopically pure KBr (1:20 w/w) and then pressed to obtain IR-transparent pellets. The FTIR was first calibrated for background signal scanning with a control sample of a pure KBr, and then the sample was scanned. The spectra were collected within a scanning range of 400-4000 cm⁻¹.

Phytotoxicity of the degraded dyes

The phytotoxicity studies were performed in laboratory at room temperature, in order to assess the toxicity of dye solution before and after biodegradation using *Triticum aestivum* and *Sorghum bicolor* seeds (10 seeds of each plate) according to Moutaouakkil et al. (2002). Each day, 2 mL of initial solution (dyes) after biodegradation (100 mgl⁻¹) was added to the seeds. The MSM solution served as a control. After 7 days, germination (%) and length of shoot and root (cm) were recorded.

Statistical analysis

Each analysis was performed using the SPSS 17.0 statistics package for Windows. The differences in germination (percentage) and length of shoot and root (cm) under different treatments were investigated with the Friedman test, followed by the Wilcoxon signed-rank test. *P* values <0.05 were considered significant.

Results

Semi-quantitative adherence assay

One adherent halotolerant *S. lentus* strain was isolated from Red Sea water displaying a high decolorization effect against three azo dyes: Congo red, Evans blue, and Eriochrome Black T (100 mgl⁻¹). The semi-quantitative adherence assay shows the ability of *S. lentus* to adhere (OD₅₇₀ = 1.2) to abiotic surface such as polystyrene plate.

Decolorization rate by S. lentus

S. lentus was tested for its ability to decolorize three azo dyes under aerobic conditions. According to Fig. 1, the maximum percentage of decolorization and the degradation time varied according to the structure of the tested dyes. Almost 100 % of Congo red and Eriochrome black T (97.63 and 96.43 %, respectively) were decolorized within 24-h incubation period, whereas the decolorization of Evans blue was low after the same time. Meanwhile, the degradation of Evans blue reached 19.7 % after 24-h incubation period. The degradation reached 50 % after 8 and 10 h for Congo red and Eriochrome black T, respectively.

Fig. 1 Decolorization of 100 mg l⁻¹ of Congo red, Evans blue, and Eriochrome black T by *Staphylococcus lentus*. [bacterium]o = 2.2×10^{6} CFU/ml, temperature = 37 °C, pH = 7.2, shaking (150 rpm)



UV–Vis characterization

Decolorization of the dyes was followed by the comparison of changes in their absorption spectrum in the range of 250–800 nm using UV–Vis spectroscopic analysis. UV–Vis spectrum of azo dyes (Congo red, Evans blue, and Eriochrome black T) is presented in Fig. 2A, B, and C, respectively.

For untreated dyes, as shown in Fig. 2A, Congo red presented one absorbance peak at 488 nm. Concerning Evans blue (Fig. 2B), the spectrum presented one absorbance peak at 611 nm. However, in Fig. 2C we noticed a large peak at 503 nm of Eriochrome black T. The treated samples after 24-h incubation period showed a decrease in optical density.

FTIR characterization

The FTIR spectra of the three studied azo dyes (Cong red, Evans blue, and Eriochrome black T) before (A) and after (B) decolorization by *S. lentus* are presented in Figs. 3, 4, and 5, respectively. Figure 3A shows that the spectra of untreated Cong red indicate a broad band stretching vibration at 3319, 2855, 1652, 1615, 1600, 1556, 1450, 1405, 1078, 1035, 713, 698, and 683 cm⁻¹.

The FTIR spectra of decolorized sample (Fig. 3B) showed a significant change in the position of peaks. Wavenumbers resulted in the degradation spectrum of Congo red showing bands at 3335, 2993, 2380, 1652, 1635, 1615, 1675, 1405, 1078, 1035, 713, 698, and 683 cm⁻¹.

Figure 4 shows FTIR spectrum of untreated and treated Evans blue dye. The spectrum indicates peaks at 3315, 2890, 2325, 1617, 1456, 1405, 1238, 1077, 1036, 915, 778, 739, and 683 cm⁻¹. FTIR spectrum of treated Evans blue

(Fig. 4B) shows a significant change in the position of peaks and disappearance of some peaks.

Figure 5 indicates FTIR spectrum of untreated and treated Eriochrome black T. The spectrum shows bands at 3343, 2925, 2850, 2359, 1633, 1417, 1230, 1077, and 1035 cm⁻¹. FTIR spectrum after microbial degradation shows a significant change in the position of peaks and disappearance of some peaks as compared to untreated dye spectrum. The spectrum of treated dye (Fig. 5B) shows disappearance of peaks at 3277, 2927, 2850, and 2359 cm⁻¹ as well as 1592, 1409, 1225, 1075, and 1033 cm⁻¹.

Phytotoxicity studies

Data presented in Table 2 showed that the mean of shoot and root lengths of *Triticum aestivum* and *Sorghum bicolor* reaches 100 % germination in MSM. While, when treated with 100 mg l⁻¹ concentration of Congo red, Evans blue, and Eriochrome Black T, the germination rate of seed was inhibited (90, 80, and 90 %, respectively). In addition, the plumule and radical lengths were affected (Table 2). Meanwhile, treating seeds with 100 mgl⁻¹ of degraded product had minor effect on the plant, as the germination percentage reaches 100 % and the shoot and root lengths grew more rapidly (P < 0.05) in comparison with nondegraded dye solution (Table 2).

Discussion

The use of *Staphylococci* for degradation of azo dyes has been reported elsewhere. *S. arlettae* causes >90 % decoloration of Reactive Yellow 107 and Reactive Red in the



Fig. 2 UV–Vis spectra of the azo dyes (100 mg) before (*straight line*) and after (*dashed line*) treatments: (**A**) Congo red, (**B**) Evans blue, and (**C**) Eriochrome black T. [bacterium]o = 2.2×10^6 CFU/mL, temperature = 37 °C, pH = 7.2, shaking (150 rpm)

presence of glucose and yeast extract (Elisangela et al. 2009).

In this study, *S. lentus* was tested for its ability to decolorize three azo dyes under aerobic conditions. It has been reported that chemical structures of azo dyes are closely related to its decolorization rate and time (Pasti-Grigsby et al. 1992). The dyes with simple structures and low molecular weights are decolorized faster than those with complex structures (Pearce et al. 2003). In our study, *S. lentus* was able to decolorize dyes that contain one (Eriochrome Black T) or two (Congo red) sulfonate groups in shorter time in comparison with dyes which had four sulfonate groups (Evans blue). Similar result was obtained with *Klebsiella* sp. strain Y3 showing a difference in decolorization rates according to the number of sulfonate groups in the tested dyes (Cui et al. 2014).

We noted also that the isolated strain was more effective on Congo red and Eriochrome black T (97.63 and 96.43 % decolorization rates, respectively) than on Evans blue (19.7 %). Similar result was obtained with *S. epidermidis*, isolated from textile wastewater plant, which has been reported to be able to cause 100 % Congo Red (750 ppm) color removal (Ayed et al. 2010). Cui et al. (2012) tested the effectiveness of azo dye removal using a consortium composed of different *Klebsiella* strains; Eriochrome black T and methyl orange were less than 80 % removed after 16 h.

According to UV–Vis spectroscopic analysis of the three azo dyes (before and after treatment), we note that the absorbance peaks in the visible region decreased without any shift in λ -max, indicating complete decolorization of the solution (Congo red and Eriochrome black T). According to Asad et al. (2007), decolorization of dyes by bacteria may be attributed to adsorption by microbial cells or to biodegradation. It has been reported that *S. arlet*-*tae* VN-11, isolated from an activated sludge process in a





Fig. 4 FTIR analysis of Evans blue (**A**) and degraded products (**B**) (after 24-h incubation period)

textile industry, was able to decolorize four different azo dyes (Reactive Yellow 107, Reactive Black 5, Reactive Red 198, and Direct Blue 71) at 100 mgl⁻¹ under microaerophilic conditions with decolorization percentage being more than 97 % (Elisangela et al. 2009). Meanwhile, a salttolerant *S. cohnii* strain, isolated from textile wastewater,

has been found to exhibit effective decolorization activity of several kinds of azo dyes with different structures at pH = 7.0 and T = 30 °C (Yan et al. 2012).

Figures 3, 4, and 5 show the FTIR spectra of the three studied azo dyes before and after decolorization. The comparison between the FTIR spectra of control dye





Table 2 Phytotoxicity study of azo dyes and its degradation product

Azo dyes	Parameters studied	Sorghum bicolor		Triticum aestivum	
		Before treatment	After biodegradation	Before treatment	After biodegradation
MSM	Germination (%)	100		100	
	Shoot (cm) \pm SD	13.5 ± 1.39		12.11 ± 1.83	
	Root (cm) \pm SD	17.06 ± 0.98		10.75 ± 0.72	
Congo red	Germination (%)	90	100	90	100
	Shoot (cm) \pm SD	6.06 ± 0.93	10.85 ± 1	4 ± 0.53	9.85 ± 1.23
	Root (cm) \pm SD	2.72 ± 0.63	6.4 ± 0.81	2.72 ± 0.42	5.8 ± 0.67
Evans blue	Germination (%)	80	100	80	100
	Shoot (cm) \pm SD	5.13 ± 0.74	8.5 ± 0.78	4 ± 0.87	9.2 ± 0.86
	Root (cm) \pm SD	2.81 ± 0.53	5.2 ± 0.54	2.5 ± 0.71	4.5 ± 0.62
Eriochrome black T	Germination (%)	90	100	90	100
	Shoot (cm) \pm SD	4 ± 1.67	8.2 ± 0.86	4.78 ± 1.20	6.55 ± 0.76
	Root (cm) \pm SD	2.3 ± 0.67	3.95 ± 0.72	2.78 ± 0.48	4.7 ± 0.64

(before degradation) with FTIR spectra of extracted metabolites after decolorization clearly indicates the biodegradation potential of parent dye compound by *S. lentus* (Fig. 3). The FTIR spectrum of treated Evans blue (Fig. 4B) and Eriochrome black T (Fig. 5B) shows a significant change in the position of peaks and disappearance of some others peaks (Sanmuga Priya et al. 2015).

A broad stretching vibrational band appeared at 3319 cm^{-1} in Fig. 3A (untreated Congo red) corresponding to N–H functional group, while saturated C-H stretching vibration band at 2855 cm⁻¹, peak at 1652 cm⁻¹ corresponding to N–H bending vibration, and N=N azo stretching vibration band at 1615 cm⁻¹. The bands at 1600 and 1556 cm⁻¹ correspond to the stretching vibrations of

aromatic ring double bonds. Moreover, the bands at 1450 and 1405 cm⁻¹ are attributed to C–H bending. The band at 1078 and 1035 cm⁻¹ are assigned to R-SO₃ groups and -S=O bands, respectively. Figure 3B indicates FTIR spectra of decolorized sample where a significant change in the position of peaks is remarked for the degradation spectrum of Congo red.

The spectrum of Evans blue (Fig. 4A) indicates a peak at 3315 cm⁻¹ assigned to stretching vibrations of O–H and N–H bond, peak at 2890 cm⁻¹ corresponding to saturated C-H stretching vibration, peak at 2325 cm⁻¹ assigned to C=O, band at 1617 cm⁻¹ corresponding to N=N azo stretching vibration, band at 1456 cm⁻¹, band at 1405 cm⁻¹ to C–H bending characteristics, band at 1238 cm⁻¹ and peaks at 1077 and 1036 cm⁻¹ are assigned to R-SO₃ groups and –S=O (Sanmuga Priya et al. 2015). Figure 4B illustrates FTIR spectrum of treated Evans blue. The broadening of the band at 3283 cm⁻¹, weakning of the peak at 2928 cm⁻¹, disappearance of the peak at 2325 cm⁻¹ and appearance of new peaks at 1583, 1453, 1405, 1238, 1079, 1038, 917, and 775 cm⁻¹ are indicative of the degradation.

FTIR spectrum (Fig. 5A) confirms the microbial degradation of Eriochrome black T where there is a significant change in the position of peaks and disappearance of some peaks as compared to untreated dye spectrum. Meanwhile, the spectrum of treated dye (Fig. 5B) shows the disappearance of peaks at 3277, 2927, 2850, and 2359 cm^{-1} as well as 1592, 1409, 1225, 1075, and 1033 cm⁻¹.

The assessment of decolorized solution on vegetation is essential to explore the possible re-use of treated water for irrigation purpose (Pourbabaee et al. 2006). The relative sensitivity of the *T. aestivum* and *S. bicolor* seeds against Congo red, Evans blue, and Eriochrome Black T (100 mgl⁻¹) before and after biodegradation is presented in Table 2. We noticed that the exposure of seeds to the untreated dye affects the germination rate (90, 80, and 90 %, respectively), while degraded solution may enhance the growth of shoot and root.

The Wilcoxon signed-rank test revealed a statistically significant difference in shoot and root lengths between the control and the treated seeds with degraded dye solution (P < 0.05).

Similar result was obtained with degraded Congo red (100 mg l^{-1}) solution by *Pseudomonas* sp. SU-EBT on *Sorghum bicolor*, *Vigna radiata*, *Lens culinaris*, and *Oryza sativa* plants as compared to parent dye (Telke et al. 2010). More recently, Saratale et al. (2015) noticed that the degraded metabolites showed non-toxic nature revealed by phytotoxicity on *Sorghum vulgare* and *Phaseolus mungo* plants. Meanwhile, Zabłocka-Godlewska et al. (2015) reported no change in toxicity of Evans Blue on *Lemna minor* after degradation by *Klebsiella* Bz4 strain.

Our study demonstrates that *S. lentus* may achieve approximately 100 % decolorization of Congo red and Eriochrome Black T (100 mg l^{-1}) by the end of treatment. UV–Visible and FTIR analyses confirmed the biodegradation potential of the tested strain. These results support the potential use of *S. lentus* for bioremediation of azo dye solution.

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